



# UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE

United States Patent and Trademark Office

Address: COMMISSIONER FOR PATENTS

P.O. Box 1450

Alexandria, Virginia 22313-1450

www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/530,694	12/28/2005	Alexander Giles Davies	35-05	9002
23713 7590 11/06/2009 GREENLEE WINNER AND SULLIVAN P C 4875 PEARL EAST CIRCLE SUITE 200 BOULDER, CO 80301				
EXAMINER				
VAN, LUAN V				
ART UNIT		PAPER NUMBER		
1795				
MAIL DATE		DELIVERY MODE		
11/06/2009		PAPER		

**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

# Office Action Summary

**Application No.**

10/530,694

**Applicant(s)**

DAVIES ET AL.

**Examiner**

LUAN V. VAN

**Art Unit**

1795

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --  
**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 22 September 2009.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1-13 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1-13 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some \* c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
  2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- 1) ☐ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO/CDC)
- 4) ☐ Interview Summary (PTO-413)
- 5) ☐ Notice of Informal Patent Application
- 6) ☐ Other: \_\_\_\_\_
- Paper No(s)/Mail Date \_\_\_\_\_

## **DETAILED ACTION**

### ***Response to Amendment***

Applicant's amendment of September 22, 2009 does not render the application allowable. Claims 1-13 are pending in the application.

### ***Status of Objections and Rejections***

All rejections from the previous office action are maintained.

### ***Claim Rejections - 35 USC § 103***

The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

Claims 1-6, 10, 12 and 13 are rejected under 35 U.S.C. 103(a) as being unpatentable over Tender et al. (*Electrochemical Patterning of the Self-Assembled Monolayers onto Microscopic Arrays of Gold Electrodes Fabricated by Laser Ablation*, Langmuir, 1996, 12, 5515-5518, cited in the IDS).

Regarding claim 1, Tender et al. teaches a method of forming coatings of at least two different coating molecules on at least two electrodes, the method comprising: (a) providing an array of at least two individually-addressable electrodes (i.e., electrodes A, Fig. 2), (b) allowing a layer of a masking molecule to adsorb onto all electrodes (i.e., immersing the substrate in ethanolic EG6S, page 5517, left column, first full paragraph); (c) inducing electrochemical desorption of the masking molecule from at least one but

not all electrodes to expose a first set of exposed electrodes (i.e. desorption of EG6S from the electrode A, page 5517, left column, first full paragraph), and (d) allowing a first coating molecule to adsorb onto the first set of exposed electrodes (i.e., immersing the electrodes in C16S, page 5517, left column, first full paragraph).

Tender et al. further teaches that the process is repeated to form more than two different monolayers as follows:

It is important to note that the extension of this technique to patterning more than two different monolayers (say n) should be straightforward. After microfabricating an array of n individually-addressable microelectrodes, exposure of the entire array to a 0.5 M ethanolic solution of an alkanethiol should result in a SAM of that alkanethiolate only on the first element if the other elements are biased to a sufficiently reductive potential or have been previously modified with another SAM. Then, by sequentially releasing potential control of the elements as the array is exposed to 0.5 M ethanolic solutions of different alkanethiols, it should be possible to build up a microscopic array consisting of n gold elements modified with SAMs of n different alkanethiolates. (Page 5517, right column, second full paragraph.)

Tender et al. differs from the instant claims in that the reference does not explicitly teach providing a masking step between application of the first coating molecule and the second coating molecule.

However, Tender et al. recognizes that "[c]ontamination of monolayers previously formed on other elements may occur by displacement of monolayer constituents by alkanethiols in solution. Such cross-contamination may be minimized, however, by using low concentrations of alkanethiols and/or using short Immersion times and/or using analogous disulfides." (See footnote 25, page 5517, left column). Furthermore, Tender et al. explains that EG6S SAMs function to resist biomolecule adsorption (i.e., to

mask biomolecule adsorption), whereas C16 SAMs function to promote biomolecule adsorption (page 5517, left column, second full paragraph).

Since Tender et al. recognizes that contamination of monolayers may occur on previously coated electrodes and that certain molecules function as a mask for adsorption, it would have been obvious to one having ordinary skill in the art at the time the invention was made to have exposed the electrodes to a masking molecule, because it would minimize the displacement of monolayer constituents by the different alkanethiols in the solution, as suggested by Tender et al. (Page 5517, right column, second full paragraph).

Regarding claim 2, Tender et al. teaches that the array can comprise  $n$  individually-addressable microelectrodes, which suggests that any number of addressable electrodes can be used. Selecting the number of individually addressable electrodes to suit the desired application would have been obvious to one having ordinary skill in the art.

Regarding claim 3, Tender et al. teaches that the extension of this technique to patterning more than two different monolayers (say  $n$ ) should be straightforward. It would have been obvious to one having ordinary skill in the art at the time the invention was made to have repeated the process multiple times in order to form a coating of more than two different monolayers, as suggested by Tender et al. (Page 5517, right column, second full paragraph).

Regarding claim 4, Tender et al. teaches that the electrode dimension is not more than 50  $\mu\text{m}$  (see Fig. 2).

Regarding claim 5, Tender et al. teaches that the separation between the electrodes is not more than 30  $\mu\text{m}$  (see Fig. 2).

Regarding claim 6, Tender et al. teaches that the electrodes are metal electrodes (i.e., gold, page 5516, right column, first full paragraph), and that the masking and coating molecules are thiolated (i.e., EG6SH and C16SH).

Regarding claim 10, Tender et al. teaches that the coating molecules are polypeptides (i.e., antibody or protein, page 5517, right column, first full paragraph) modified with a function of group capable of adsorbing onto the electrodes

Regarding claim 12, since Tender et al. teaches that the technique can be extended to patterning more than two different monolayers with an array of  $n$  individually-addressable electrodes, it would have been obvious to one having ordinary skill in the art at the time the invention was made to have controlled the potential of electrodes from which desorption is not required, because it would prevent the previously formed coating on the electrodes from being desorbed. Since the electrodes are individually addressable, they can be individually controlled.

Regarding claim 13, the application of the voltage by Tender et al. is either AC or DC, since an electrical potential can only be applied by either AC and DC.

Claims 7-9 are rejected under 35 U.S.C. 103(a) as being unpatentable over Tender et al. in view of Barton et al. (WO 99/51778).

Tender et al. teaches the method as described above. Tender et al. differs from the instant claims in that the reference does not explicitly teach the specific coating molecules of the instant claims.

Barton et al. teaches a highly sensitive and accurate method for the detection of genetic point mutations in nucleic acid sequences and its application as a biosensor. In particular, the invention relates to electrodes that are prepared by modifying their surfaces with oligonucleotide duplexes combined with an intercalative, redox-active species and their use as sensors based on an electrochemical process in which electrons are transferred between the electrode and the redox-active species (page 8, lines 25-31).

It would have been obvious to one having ordinary skill in the art at the time the invention was made to have deposited the oligonucleotides of Barton et al. in the method of Tender et al., because it would enable the electrodes to function as a biosensor for the detection of genetic mutations in the nucleic acid sequences (page 8, lines 25-31 of Barton et al.).

Claim 11 is rejected under 35 U.S.C. 103(a) as being unpatentable over Tender et al. in view of Chan (US patent 6355420).

Tender et al. teaches the method as described above. Tender et al. differs from the instant claims in that the reference does not explicitly teach applying an electric field.

Chan teaches that the orientation of DNA in an electric field has been well studied (column 84 lines 40-43), and that DNA molecules and other polymers align themselves in the direction of electric fields whether in an electrophoretic gel or in solution (column 85 lines 19-23). The implications of DNA alignment in an electric field further support the fact that DNA molecules and other polymers can be driven across nanochannels in a linear fashion (column 85 lines 23-26).

It would have been obvious to one having ordinary skill in the art at the time the invention was made to have applied an electric field, as taught by Chan, in the method of Tender et al., because it would align the coating molecules, such as DNA molecules or other polymers, in the direction of the electric field (column 85 lines 19-23 of Chan). The application of the electric field is either AC or DC, since an electrical potential for providing the electric field can only be applied by either AC or DC.

### ***Response to Arguments***

Applicant's arguments filed have been fully considered but they are not persuasive. In the arguments presented on page 7-8 of the amendment, the applicant argues that Tender et al. does not contemplate a re-protection step. The examiner respectfully disagrees. Since Tender et al. teaches that that EG6S SAMs function to resist biomolecule adsorption, i.e., to mask biomolecule adsorption, (page 5517, left column, second full paragraph) and that the "technique to patterning more than two different monolayers (say n) should be straightforward" (page 5517, right column, second full paragraph), one having ordinary skill in the art would expect that by



repeating the patterning step, i.e., repeating the deposition of EG6S, the masking step is repeated between the coating steps in forming the C16S SAMs.

### ***Conclusion***

**THIS ACTION IS MADE FINAL.** See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire **THREE MONTHS** from the mailing date of this action. In the event a first reply is filed within **TWO MONTHS** of the mailing date of this final action and the advisory action is not mailed until after the end of the **THREE-MONTH** shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than **SIX MONTHS** from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to LUAN V. VAN whose telephone number is (571)272-8521. The examiner can normally be reached on M-F 9:30-6:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Nam Nguyen can be reached on 571-272-1342. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Nam X Nguyen/  
Supervisory Patent Examiner, Art Unit 1753

LVV  
November 4, 2009